

PUBLICATION NUMBER : 10306208
PUBLICATION DATE : 17-11-98

APPLICATION DATE : 08-05-97
APPLICATION NUMBER : 09117963

APPLICANT : MITSUBISHI ENG PLAST KK;

INVENTOR : NAKADA MICHIO;

INT.CL. : C08L 69/00 C08K 5/42 C08K 5/521 //(C08L 69/00 , C08L 27:18 , C08L 33:10)

TITLE : POLYCARBONATE RESIN COMPOSITION

ABSTRACT : PROBLEM TO BE SOLVED: To obtain a polycarbonate resin composition, excellent in flame retardance and thermal stability and having a high impact strength.

SOLUTION: This polycarbonate resin composition is obtained by compounding 100 pts.wt. aromatic polycarbonate resin having 16,000-30,000 viscosity-average molecular weight with 0.5-40 pts.wt. phosphorus-based flame retardant, 0.01-5 pts.wt. polyfluoroethylene resin, 0.01-5 pts.wt. metal sulfonate and 0.5-30 pts.wt. multilayered structural polymer having an outermost shell layer composed of an alkyl (meth)acrylate-based polymer.

COPYRIGHT: (C)1998,JPO

(19)



Europäisches Patentamt
European Patent Office
Office européen des brevets



(11)

EP 0 289 828 B1

(12)

EUROPEAN PATENT SPECIFICATION

(45) Date of publication and mention
of the grant of the patent:
21.08.1996 Bulletin 1996/34

(51) Int Cl.⁶: **C08L 69/00, C08L 9/00,
C08L 51/04, C08K 5/43,
C08L 27/18, C08K 5/02,
C08K 5/49, C08K 5/41**

(21) Application number: **88105969.5**

(22) Date of filing: **14.04.1988**

(54) **Fire retardant impact modified carbonate polymer composition and process for preparing the same**

Feuerhemmende, schlagmodifizierte Karbonatpolymerzusammensetzung und Verfahren zu deren Herstellung

Composition de polycarbonate ignifugée à résistance au choc modifiée et son procédé de préparation

(84) Designated Contracting States:
AT BE CH DE FR GB IT LI NL SE

(30) Priority: **06.05.1987 US 46364**

(43) Date of publication of application:
09.11.1988 Bulletin 1988/45

(73) Proprietor: **THE DOW CHEMICAL COMPANY**
Midland Michigan 48640-1967 (US)

(72) Inventors:
• **Laughner, Michael K.**
Lake Jackson Texas 77566 (US)
• **Ogoe, Samuel A.**
Angleton Texas 77515 (US)

(74) Representative: **Casalonga, Axel et al**
BUREAU D.A. CASALONGA - JOSSE
Morassistrasse 8
80469 München (DE)

(56) References cited:

EP-A- 182 226	EP-A- 0 019 127
EP-A- 0 162 243	EP-A- 0 173 358
US-A- 3 162 695	US-A- 4 239 678
US-A- 4 360 609	US-A- 4 626 563

- **Encyclopedia of Polymer Science and Technology, Vol 1; pages 397-398**
- **Römpp Chemie Lexikon, 9.Ed., page 3107**
- **Hawley's Condensed Chemical Dictionary, 11th edition, page 855**

Remarks:

The file contains technical information submitted after the application was filed and not included in this specification

Note: Within nine months from the publication of the mention of the grant of the European patent, any person may give notice to the European Patent Office of opposition to the European patent granted. Notice of opposition shall be filed in a written reasoned statement. It shall not be deemed to have been filed until the opposition fee has been paid. (Art. 99(1) European Patent Convention).

EP 0 289 828 B1

Description

This invention relates to impact modified carbonate polymer compositions containing additives which act to reduce the susceptibility of such polymer to burning or at least to minimize the effects of dripping of molten polymer when exposed to fire conditions.

Carbonate polymers derived from reactions of dihydroxyorganic compounds, particularly the dihydric phenols, and carbonic acid derivatives such as phosgene, have found extensive commercial application because of their excellent physical properties. These thermoplastic polymers are suitable for the manufacture of molded parts wherein impact strength, rigidity, toughness, heat resistance, and excellent electrical properties are required.

Unfortunately, however, these polymers exhibit a brief but definite burning time when contacted with an open flame. More importantly, as is often the case, the carbonate polymers contain stabilizers and other additives which are often more combustible than the unmodified carbonate polymer. This is particularly true in the case of impact modified polycarbonates wherein the presence of as little as one percent of a rubbery polymer to improve the impact properties of the blend can lead to substantially increased burn times and dripping of flaming material. As a result, the modified carbonate polymers frequently exhibit substantially poorer resistance to combustion than do the unmodified carbonate polymers.

In attempts to increase the combustion resistance of carbonate polymers, including the impact modified forms thereof, it has been a common practice to incorporate monomeric phosphites, phosphoric acid esters, thiophosphoric acid esters containing halogenated alkyl radicals and halogenated organic compounds into the carbonate polymer. However, in order to obtain any noticeable improvement in combustion resistance, these additives have been employed in such large quantities that they often adversely affect many of the desirable physical and mechanical properties of the carbonate polymer.

In US-A- 4,463,130 there were disclosed certain blends of a polycarbonate resin, a styrene polymer and a branched polyphosphate. In addition the composition may include polytetrafluoroethylene. Included within the term "styrene polymer" were styrene graft polymers of rubbery substrates including diene, acrylate, or EPDM (ethylene, propylene, diene monomer) rubbers. Examples 2-5 disclosed blends containing polybutadiene modified polystyrene (HIPS), polycarbonate, polytetrafluoroethylene and additional components.

In US-A- 4,617,345 a polymer mixture comprising an aromatic polycarbonate, a partially hydrogenated block copolymer and a rubbery graft copolymer is disclosed. Fire retarding additives could also be present.

In US-A- 4,299,928 acrylate rubbers were employed in polycarbonate compositions also including fire retarding additives.

In US-A- 4,626,928 fire retardant polycarbonates comprising metal salts of sulfimides, chlorinated or brominated bisphenol polycarbonates and polytetrafluoroethylene were disclosed.

US-A- 3 162 695 discloses blends of polycarbonates with graft copolymers containing butadiene, methyl methacrylate and styrene. The amount of the graft copolymers in these blends is rather high and may vary from 20 to 95 wt%. The patent does not touch upon the feature of combustion resistance or fire retardance of the blends.

EP-A- 182 226 discloses a flame-resistant moulding composition, comprising a polycarbonate, a rubbery polymer and a tetrafluoroethylene polymer. Suitable rubbers may be graft polymers and are listed in col.5, lines 19-65. Particularly preferred graft polymers are said to be the so-called ABS-polymers.

The present invention is an impact resistant, fire retardant carbonate polymer composition comprising a mixture of:

- a) one or more carbonate polymers,
- b) from 0.1 to 10 percent by weight of a rubbery polymer based on total composition weight, which rubbery polymer has a $T_g < 20^\circ \text{C}$, is unsaturated and contains, on average, at least one ethylenically unsaturated group per 25 carbon atoms of the rubbery polymer, and which rubbery polymer is a graft of styrene/methacrylate copolymer on a butadiene homopolymer or copolymer core, and
- c) 0.01 to 10 weight percent based on total composition weight of a fibril-forming polytetrafluoroethylene.

Hereinafter, such compositions will be referred to as fire retardant carbonate polymer compositions. These compositions exhibit surprisingly high resistance to combustion as well as physical properties comparable to an impact modified carbonate polymer containing no fire retardant additive.

In preferred embodiments, the invention additionally comprises one or more components such as metal salts of aromatic sulfur-containing compounds, monomeric or polymeric chloro- or bromo-organic compounds, organophosphorous compounds capable of providing improved polymer heat stability, and processing stability enhancers such as the metal salts of inorganic acids, especially sodium bisulfate.

In a most preferred embodiment, the compositions are substantially free of ungrafted styrene polymer or styrene-containing copolymers. The presence of such ungrafted polymers or copolymers in the matrix leads to inferior fire retardant properties and accordingly is undesired.

The fire retardant impact modified carbonate polymer compositions of the present invention are suitably employed in most applications in which opaque impact modified polycarbonates have been previously utilized. Applications of particular interest for the utilization of the said carbonate polymer compositions of this invention are pigmented and/or colored carbonate polymers useful as: automobile parts, e.g., air filters, fan housings, exterior components, housings for electrical motors, appliances, business and office equipment, photographic equipment, and aircraft applications.

The carbonate polymers employed in the present invention are advantageously aromatic carbonate polymers, such as the trityl diol carbonates described in US-A- Nos. 3,036,036; 3,036,037; 3,036,038 and 3,036,039; polycarbonates of bis(ar-hydroxyphenyl)-alkylidenes (often called bisphenol-A type diols), including their aromatically and aliphatically substituted derivatives such as disclosed in U.S. Patent Nos. 2,999,835; 3,038,365, 3,334,154, and 4,299,928; and carbonate polymers derived from other aromatic diols such as described in U.S. Patent No. 3,169,121.

It is understood, of course, that the carbonate polymer may be derived from (1) two or more different dihydric phenols or (2) one or more dihydric phenols and one or more hydroxy- or acid-terminated reactants such as dicarboxylic acids, or alkylene glycols in the event a carbonate copolymer or interpolymer rather than a homopolymer is desired. Also suitable for the practice of this invention are blends of any one of the above carbonate polymers. Also included in the term "carbonate polymer" are the ester-carbonate copolymers of the types described in U.S. Patent Nos. 3,169,121; 4,287,787; 4,156,069; 4,260,731 and 4,105,633. Of the aforementioned carbonate polymers, the polycarbonates of bisphenol-A and derivatives, including copolycarbonates of bisphenol-A, are preferred. Methods for preparing carbonate polymers for use in the practice of this invention are well known; for example, several suitable methods are disclosed in the aforementioned patents.

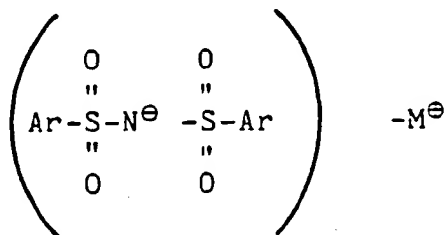
The rubbery polymers containing residual ethylenic unsaturation have been found uniquely suited to provide impact resistance in an ignition resistant carbonate polymer composition without adversely affecting the burning characteristics of such polymer. In contrast, highly saturated rubbery polymers including those previously suggested for use in the impact modification of non ignition-resistant carbonate polymers such as, for example, acrylate rubbers and hydrogenated polydiolefin based rubbers do not provide the advantageous resistance to the effects of burning achieved by the present invention. This result is believed to be due to the ability of specified rubber to crosslink under burning conditions and in the presence of the remaining components of the resin blend. This crosslinking ability substantially reduces or eliminates the tendency of the impact modified carbonate polymer blend to form increased amounts of char and dripping of molten polymer.

The suitable rubbery polymers are those polymers having a Tg less than 20°C, and sufficient residual unsaturation to result in improved performance under burning conditions compared to similar carbonate polymer formulations utilizing saturated rubbers as measured for example by the UL-94 test. The unsaturated rubbers employed in the invented compositions contain, on average, at least one ethylenically unsaturated group per 25 carbons of the rubbery polymer, most preferably at least one ethylenically unsaturated group per 10 carbons of the rubbery polymer. The rubbery polymers are graft polymers of the shell/core type comprising a graft of styrene/ methyl methacrylate copolymer on a butadiene homopolymer or copolymer core. Such compositions have been previously disclosed in US-A- 4,617,345. Examples of the latter class of graft polymers include those compositions available commercially under the trade name Paraloid® 3607 available from Rohm and Haas Company.

The suitable fluorine-containing polymer is a polytetrafluoroethylene adapted to form a fibrillar structure to stabilize the polymer under molten conditions, as disclosed by US-A- 3,005,795, 3,671,487 and 4,463,130.

Additional components may be employed in the present composition to achieve even better fire retardant and other polymer properties. Such additional components include metal salts of sulfur compounds, and chlorine or bromine containing compounds.

The metal salts of sulfur compounds are metal salts of aromatic sulfonates, sulfates, sulfonamides and sulfimides. A preferred aromatic sulfur compound is an aromatic sulfimide represented by the formula:



wherein Ar is an aromatic or substituted aromatic group and M is a monovalent metal cation. Suitable monovalent metals include Group I metals as well as copper. M is preferably an alkali metal, most preferably sodium or potassium.

Representative preferred sulfimide salts include the alkali metal salts of saccharin, N-(p-tolylsulfonyl)-p-toluene-sulfimide, N-(N'-benzylaminocarbonyl)sulfanilimide, N-(phenylcarboxyl)-sulfanilimide, N-(2-pyrimidinyl)-sulfanilimide,

N-(2-thiazolyl)sulfanilimide and other monovalent metal salts of the sulfimides disclosed in U.S. Patent No. 4,254,015. Combinations of the above-identified salts can also be employed.

Useful monomeric chloro- or bromo-halogenated organic compounds are halogenated diphenyl sulfone, diphenyl, diphenyl oxide, benzene, phthalimide, and mixtures thereof. These compounds can be mono- or poly-halogenated preferably with bromine atoms since they cause less degradation of the polycarbonate and appear to be more efficient as flame retardants.

Useful polymeric chloro- or bromo-halogenated organic compounds are polymeric carbonate derivatives of a chlorinated or brominated bisphenol with or without an additional nonhalogenated bisphenol. Thus, homopolymers and copolymers are useful in this invention. In addition, the polymeric chloro- or bromo-halogenated compound may be a halogenated polymer other than a polycarbonate. Examples include polymers of halogenated vinyl aromatic monomers such as chlorinated or brominated styrene containing polymers, taught for example in U.S. Patent 3,896,185.

As previously mentioned, the invented composition advantageously employs an organophosphorous compound to aid in preventing polymer degradation under molding conditions. Suitable organo phosphorous compounds are the phosphorous esters such as those disclosed in U.S. Patent 4,514,357, and references cited during the prosecution thereof.

Useful melt stabilizing additives are metal bisulfates or sulfates having a pka of from 1 to 5 and preferably 1 to 3. Examples of the latter include sodium bisulfate, potassium bisulfate, lithium bisulfate, rubidium bisulfate, cesium bisulfate, and aluminum sulfate.

The carbonate polymer compositions of the present invention are suitably prepared by combining the ingredients in effective amounts using any of a variety of blending procedures conventionally employed for polymer blends. For example, dry particulates of the carbonate polymer, rubbery polymer, and the other additives can be dry blended and the resulting dry blend extruded into the desired shape. By "effective amount" is meant the relative amount of the respective components that is sufficient to provide the desired polymer characteristic.

The rubbery polymer is employed in an amount to result in improved impact strengths of the resulting blend, that is from 0.1 to 10 wt. percent based on total composition weight.

The metal salt of an aromatic sulfur compound, monomeric or polymeric chloro- or bromo-organic compound, and fluorine-containing polymer are employed in respective amounts to provide the desired degree of resistance to the effects of burning. For example, a composition may be formulated to provide the desired "V" rating when tested under the conditions of the UL-94 test. Thus the metal salt of an aromatic sulfurcontaining compound is employed in an amount that imparts to the composition a desired level of fire retardancy. Preferred amounts are in the range of from 0.001 to 2 percent, especially from 0.001 to 1.0 weight percent, based on total composition weight.

Similarly, the amount of the chloro- or bromo-halogenated organic compound is selected to provide a final halogen content of composition in the range of from 0.01 to 10 percent by weight and most preferably from 0.1 to 1.0 percent by weight. It is to be understood that one can use relatively small amounts of a highly halogenated aromatic compound or a relatively large amount of a sparsely halogenated compound to achieve the above desired ranges.

The amount of the fibril-forming tetrafluoroethylene polymer is in the range of from 0.01 to 10 percent and preferably in the range of from 0.1 to 5 percent by weight based on total composition weight.

Where employed, the additional components, organophosphorous compounds and metal salts of inorganic acids are suitably present in sufficient amounts to impart resistance to degradation in the melt, i.e., lack of color formation or molecular weight degradation under molding conditions, and melt stability. Preferred amounts are from 0.005 to 1 percent by weight of organophosphorous compound, and from 0.001 to 1 percent by weight of metal salt of an inorganic acid.

In addition, other additives can be included in the impact modified carbonate polymer composition of the present invention such as fillers (i.e., glass fibers), pigments, dyes, antioxidants, heat stabilizers, ultraviolet light absorbers, mold release agents and other additives commonly employed in carbonate polymer compositions.

The following Examples and Comparative Runs are given to further illustrate the invention and should not be construed as limiting its scope. In the following examples, all parts and percentages are by weight unless otherwise indicated.

Examples 1 to 8 and Comparative Runs A to E

Carbonate polymer formulations containing a bisphenol A polycarbonate (CALIBRE® 300) available from The Dow Chemical Company, having a molecular weight of 22,000 (BPA/PC), polytetrafluoroethylene (PTFE), various rubbers, and optional components: tetrabromobisphenol A polycarbonate oligomer (BC-52) available from Great Lakes Chemicals, Inc. (TBBPA/PC); potassium paratolylsulfimide (KpTSM) and potassium bisulfate (KHSO₄) were prepared containing respective amounts of the ingredients as indicated in Table I. Also prepared were control compositions of only BPA/PC and a composition lacking rubber. The unsaturated rubber tested was a methyl methacrylate/ styrene copolymer grafted polybutadiene (Paraloid® 3607) available from Rohm and Haas. Comparative examples included a meth-

acrylate/styrene grafted butyl acrylate (Paraloid® 3330), a styrene/acrylonitrile grafted EPDM (VISTALON® 3708) available from Exxon Corporation, and a hydrogenated styrene/butadiene block copolymer (KRATON G-1651) available from Shell.

Results of testing various properties including resistance to burning according to UL-94 are contained in Table I.

Table I

Examples and Comparative Runs	Composition (pt. by wt.)	Izod Impact* (J/mm notch)	% Brittle**	UL-94	Burn Time (sec.)	# Drips
A	100 BPA/PC	16.2 (0.865)	0	V-2	20.0	5.5
B	98.5 BPA/PC 0.1 KpTSM 0.1 KHSO ₄ 1.0 TBBA/PC 0.3 PTFE	2.5 (0.133)	100	V-0	1.0	0/5
C	98.2 BPA/PC 0.057 KpTSM 0.05 KHSO ₄ 0.5 TBBA 0.2 PTFE 1.0 PARALOID 3330	7.2 (0.384)	38	V-2	9.0	1/5
D	98.2 BPA/PC 0.05 KPTSM 0.05 KHSO ₄ 0.5 TBBA 0.2 PTFE 1.0 VISTALON 3708	12.4 (0.662)	13	V-2	7.0	5/5
E	98.2 BPA/PC 0.05 KPTSM 0.05 KHSO ₄ 0.5 TBBA 0.2 PTFE 1.0 KRATON G-1651	10.0 (0.534)	25	V-2	20.1	5/5
1	98.2 BPA/PC 0.05 KPTSM 0.05 KHSO ₄ 0.5 TBBA 0.2 PTFE 1.0 PARALOID 3607	15.2 (0.811)	0	V-0	1.2	0/5

Table I (Cont.)

5

10	Examples and Comparative Runs	Composition (pt. by wt.)	Izod Impact* (J/mm notch)	% Brittle**	UL-94	Burn Time (sec.)	# Drips
15	2	96.5 BPA/PC 0.1 KPTSM 0.1 KHSO ₄ 1.0 TBBA 0.3 PTFE 1.0 PARALOID 3607	14.8 (0.79)	0	V-0	1.0	0/5
20	3	96.5 BPA/PC 0.1 KPTSM 0.1 KHSO ₄ 1.0 TBBA 0.3 PTFE 2.0 PARALOID 3607	15.0 (0.80)	0	V-0	1.1	0/5
25	4	94.5 BPA/PC 0.1 KPTSM 0.1 KHSO ₄ 1.0 TBBA 0.3 PTFE 3.0 PARALOID 3607	15.2 (0.81)	0	V-0	1.4	0/5

30

35

40

45

50

55

Table I (Cont.)

5

10	Examples and Comparative Runs	Composition (pt. by wt.)	Izod Impact* (J/mm notch)	% Brittle**	UL-94	Burn Time (sec.)	# Drips
	5	98.6 BPA/PC 0.1 KHSO ₄ 0.3 PTFE 1.0 PARALOID 3607	14.9 (0.80)	0	V-0	0.9	0/5
15							
20	6	96.6 BPA/PC 0.1 KHSO ₄ 0.3 PTFE 3.0 PARALOID 3607	15.0 (0.80)	0	V-0	1.3	0/5
25	7	98.7 BPA/PC 0.3 PTFE 1.0 PARALOID 3607	15.2 (0.81)	0	V-0	1.5	0/5
30	8	96.7 BPA/PC 0.3 PTFE 3.0 PARALOID 3607	15.5 (0.83)	0	V-0	1.8	0/5

35

40

* 10-mil (0.25 millimeter) Izod impact strength 40°F (4.4°C)
(ft. lbs/in. notch).

45

** Percentages of samples showing brittle failure
(ASTM D-256).

As may be seen by reference to Table I, good impact properties and fire resistant properties are provided in the present invention.

50

Claims

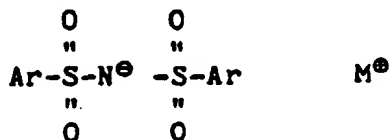
1. An impact resistant, fire retardant carbonate polymer composition comprising a mixture of

55

- a) one or more carbonate polymers,
- b) from 0.1 to 10 percent by weight of a rubbery polymer based on total composition weight, which rubbery polymer has a Tg < 20° C, is unsaturated and contains, on average, at least one ethylenically unsaturated

group per 25 carbon atoms of the rubbery polymer, and which rubbery polymer is a graft of styrene/methacrylate copolymer on a butadiene homopolymer or copolymer core, and
c) 0.01 to 10 weight percent based on total composition weight of a fibril-forming polytetrafluoroethylene.

2. An impact resistant carbonate polymer composition as claimed in Claim 1 additionally comprising from 0.001 to 2 wt% based on total composition weight of a metal salt of an aromatic sulfonate, aromatic sulfate, aromatic sulfonamide or aromatic sulfimide.
3. A composition as claimed in Claim 2 wherein the metal salt is a compound represented by the formula:

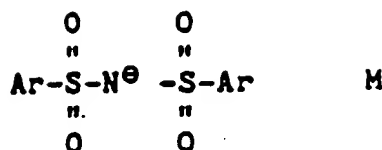


wherein Ar is an aromatic or substituted aromatic group and M is a monovalent metal cation.

4. A polycarbonate composition according to claim 1, having a halogen content in the range of 0.01 to 10 percent by weight and comprising:
a chlorine or bromine containing compound selected from monomeric chlorinated or brominated organic compounds and polymeric chlorinated or brominated organic compounds.
5. A polycarbonate composition according to claim 1 or 4, having a halogen content in the range of 0.01 to 10 percent by weight and further comprising:
- from 0.005 to 1 percent by weight based on the total composition weight of an organophosphorous compound.
6. A composition as claimed in Claim 4 or 5, wherein the chlorine- or bromine-containing compound is an oligomeric or polymeric carbonate derivative of a chlorinated or brominated bisphenol.
7. A composition as claimed in any one of the preceding claims wherein the carbonate polymer is a polycarbonate of a bis(ar-hydroxyphenyl) alkylidene or other aromatic diol.
8. A composition as claimed in Claim 7 wherein the carbonate polymer is a polycarbonate of bisphenol A.

Patentansprüche

1. Schlagfeste feuerhemmende Carbonatpolymerzusammensetzung, umfassend eine Mischung von:
 - a) einem oder mehreren Carbonatpolymeren,
 - b) von 0,1 bis 10 Gew.-% eines kautschukartigen Polymeren, bezogen auf Gewicht der Gesamtzusammensetzung, wobei dieses kautschukartige Polymere einen Tg-Wert $< 20^{\circ}\text{C}$ hat, ungesättigt ist und im Durchschnitt wenigstens eine ethylenartig ungesättigte Gruppe auf 25 Kohlenstoffatome des kautschukartigen Polymeren enthält und wobei dieses kautschukartige Polymere eine Pfropfung von Styrol/Methacrylatcopolymerem auf einem Butadienhomopolymeren- oder -copolymerenkern ist, und
 - c) 0,01 bis 10 Gew.-%, bezogen auf Gewicht der Gesamtzusammensetzung, eines fibrillenbildenden Polytetrafluorethylens.
2. Schlagfeste Carbonatpolymerzusammensetzung nach Anspruch 1, zusätzlich umfassend von 0,001 bis 2 Gew.-%, bezogen auf Gewicht der Gesamtzusammensetzung, eines Metallsalzes eines aromatischen Sulfonates, aromatischen Sulfates, aromatischen Sulfonamids oder aromatischen Sulfimids.
3. Zusammensetzung nach Anspruch 2, worin das Metallsalz eine durch die folgende Formel wiedergegebene Verbindung ist:

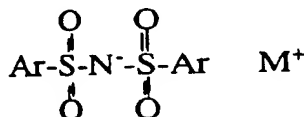


worin Ar eine aromatische oder substituierte aromatische Gruppe ist und M ein einwertiges Metallkation ist.

4. Polycarbonatzusammensetzung nach Anspruch 1, welche einen Halogengehalt in dem Bereich von 0,01 bis 10 Gew.-% aufweist und umfaßt: eine chlor- oder bromhaltige Verbindung, ausgewählt aus monomeren chlorierten oder bromierten organischen Verbindungen und polymeren chlorierten oder bromierten organischen Verbindungen.
5. Polycarbonatzusammensetzung nach Anspruch 1 oder 4, welche einen Halogengehalt in dem Bereich von 0,01 bis 10 Gew.-% aufweist und weiter umfaßt: von 0,005 bis 1 Gew.-%, bezogen auf Gewicht der Gesamtzusammensetzung, einer organophosphorverbindung.
6. Zusammensetzung nach Anspruch 4 oder 5, worin die chlor- oder bromhaltige Verbindung ein polymeres Carbonatderivat eines chlorierten oder bromierten Bisphenols ist.
7. Zusammensetzung nach einem der vorhergehenden Ansprüche, worin das Carbonatpolymere ein Polycarbonat eines Bis-(ar-hydroxyphenyl)-alkylidens oder anderen aromatischen Diols ist.
8. Zusammensetzung nach Anspruch 7, worin das Carbonatpolymere ein Polycarbonat von Bisphenol-A ist.

Revendications

1. Composition de polycarbonate ignifugée et résistante au choc, qui contient un mélange constitué
 - a) d'un ou de plusieurs polycarbonates,
 - b) de 0,1 à 10 % en poids, par rapport au poids total de la composition, d'un polymère caoutchouteux dont la température de transition vitreuse T_v est inférieure à 20°C et qui contient en moyenne au moins 1 groupe à insaturation éthylénique pour 25 atomes de carbone de ce polymère caoutchouteux, lequel est un produit résultant du greffage d'un copolymère styrène/méthacrylate sur un noyau d'homopolymère ou de copolymère de butadiène, et
 - c) de 0,01 à 10 % en poids, par rapport au poids total de la composition, d'un polytétrafluoroéthylène fibrilligène.
2. Composition de polycarbonate résistante au choc, conforme à la revendication 1, qui contient en outre de 0,001 à 2 % en poids, par rapport au poids total de la composition, d'un sel métallique d'un sulfonate aromatique, d'un sulfate aromatique, d'un sulfonamide aromatique ou d'un sulfimide aromatique.
3. Composition conforme à la revendication 2, dans laquelle le sel métallique est un composé représenté par la formule



dans laquelle Ar représente un groupe aromatique portant ou non des substituants et M représente un cation de métal monovalent.

4. Composition de polycarbonate conforme à la revendication 1, dont la teneur en halogène vaut de 0,01 à 10 % en poids et qui contient un composé chloré ou bromé, choisi parmi les composés organiques monomères chlorés ou bromés et les composés organiques polymères chlorés ou bromés.

5. Composition de polycarbonate conforme à la revendication 1 ou 4, dont la teneur en halogène vaut de 0.01 à 10 % en poids et qui contient en outre de 0,005 à 1 % en poids, par rapport au poids total de la composition, d'un composé organophosphoré.
- 5 6. Composition conforme à la revendication 4 ou 5, dans laquelle le composé chloré ou bromé est un polycarbonate dérivé d'un bisphénol chloré ou bromé.
7. Composition conforme à l'une des revendications précédentes, dans laquelle le polycarbonate est un polycarbonate dérivé d'un bis(hydroxyphényl)alkylidène ou d'un autre diol aromatique.
- 10 8. Composition conforme à la revendication 7, dans laquelle le polycarbonate est un polycarbonate dérivé du bisphénol A.

15

20

25

30

35

40

45

50

55